

Electronic structure of infinite-layer SrFeO₂ film prepared via topotactic reduction

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In this research, we tried to investigate electronic structure of infinite-layer SrFeO₂ single crystalline film by using angle-resolved photoemission spectroscopy (ARPES). Because ARPES measurement requires clean surface of samples, we performed surface cleaning of the film via ultra-high vacuum annealing method. The annealing resulted in partially clean surface of the SrFeO₂ single crystalline film, however, a clear ARPES image was still difficult to be obtained.

0. About the authors

T. Katayama : He specializes in synthesis of anion doped oxide thin films with different anion orientations via pulsed laser deposition (PLD) and topotactic reactions. In this study, he synthesized thin-film samples by combining PLD and low-temperature topotactic reduction using CaH₂.

T. Sonobe : T. S. specializes in the angle-resolved photoemission spectroscopy (ARPES) study on the electronic structure of the iron-based superconductors. His research interest is on the electronic structure of SrFeO₂ including the square planar Fe²⁺ layer like the iron-based superconductors. He performed the annealing process for surface cleaning and ARPES.

1. Introduction

Transition metal oxides are widely studied due to their intriguing features such as photocatalysis, ferroelectric, and colossal magnetoresistance. The properties of the oxides are strongly associated with the coordination of transition metal centers. Recently, low-temperature topotactic technique using metal hydrides enables to fabricate unique coordinations. For example, SrFeO₂, which is obtained by heating SrFeO_{3-δ} with CaH₂, possesses FeO₄ square-planer coordination as shown in Fig. 1, in contrast to conventional iron oxides which have three dimensionally FeO₆ octahedral, FeO₅ pyramidal and/or FeO₄ octahedral coordinations [1]. On the other hand, heating SrVO₃ with CaH₂ yields oxyhydride SrVO₂H with VO₄H₂ octahedral coordination, where hydride ions locate at apical sites [2].

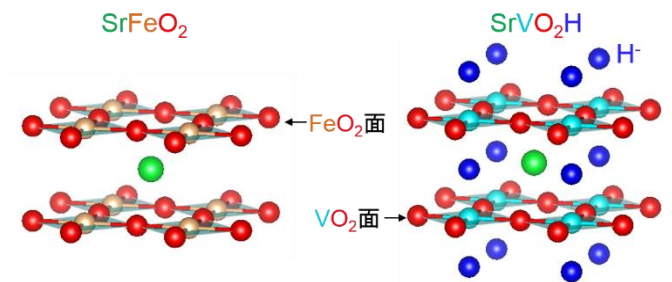


Fig. 1. Crystal structure of SrFeO₂ and SrVO₂H.

SrFeO₂ has attracted great interest because of the strong antiferromagnetic interaction with high Néel temperature of 473 K [1] and high electron mobility in a form of thin film [3]. So far, the electronic density states and electron configuration of iron in SrFeO₂ have been investigated by both theoretical calculation [4] and experimental X-ray photoemission and absorption spectroscopy methods [5]. However, the electronic structure has not been completely studied, and the origin of the high electron mobility is still unclear. Thus, we attempted further investigation of electronic structure of SrFeO₂ by using ARPES.

The ARPES measurement requires clean surface of single crystalline sample. In SrFeO₂, the single crystalline phase is only obtained in a form of film [6]. Therefore, we tried to perform ARPES measurement on single crystalline SrFeO₂ film, where its surface is cleaned through ultra-high vacuum annealing method.

2. Experiment

This work contains two steps; firstly, we checked crystal structure stability of SrFeO₂ films against ultra-high vacuum

annealing by using X-ray diffraction (XRD) measurements, and then we performed ARPES on the films.

The SrFeO₂ epitaxial thin films with infinite-layer structure were prepared on 0.5 wt% Nb-doped STO (Nb:STO) substrates by combining the pulsed laser deposition (PLD) method with topotactic reduction using CaH₂, as reported in Ref. 3. The conductive Nb:STO substrates were used so as to reduce the charge-up effect. The thickness of the films was set as ~50 nm. The crystal structure stability of the SrFeO₂ films were examined by annealing at 220 and 400°C under vacuum pressure of 1×10^{-7} Torr. Crystal structures of the films were evaluated by using XRD with Cu K α radiation.

For ARPES measurements, we used ARPES apparatus with R4000 analyser and light from a helium discharge tube (HeII α = 40.814 eV). In this apparatus, samples can be introduced in ultra-high vacuum pressure after annealing process ($\leq 1 \times 10^{-8}$ Torr). We performed ARPES measurements for the SrFeO₂ film and films obtained by annealing the SrFeO₂ films at 200 and 400°C for 2 h in ultra-high vacuum pressure. The all measurements were conducted at 300 K.

3. Results and discussion

3.1. Crystal structure

Figure 2 shows XRD patterns of the SrFeO₂ film and films obtained by annealing the SrFeO₂ film at 220 and 400°C for 2 h under vacuum pressure of 1×10^{-7} Torr. The SrFeO₂ film shows 002 XRD diffraction peak, probing that the film has *c*-axis infinite-layer structure. After annealing at 220°C, the 002 peak position did not shift, indicating that the single crystalline SrFeO₂ film has crystal structure stability against ultra-high vacuum annealing at 220°C. On the other hand, after annealing at 400°C, the 002 peak intensity decreased a little.

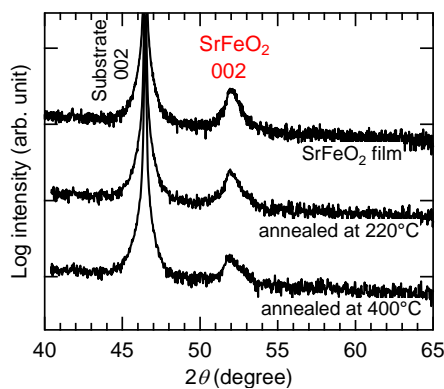


Fig. 2. XRD patterns for the SrFeO₂ film and films obtained by annealing the SrFeO₂ film at 220 and 400°C for 2 h under vacuum pressure of 1×10^{-7} Torr.

3.2. Attempt of electronic structure investigation of the SrFeO₂ films

ARPES measurements were conducted for the films. Figure 3(a) plots ARPES image of the SrFeO₂ film before annealing. Though some bands show 1eV-order band width in the band calculation[4], the spectrum does not show any angle dependence. It indicates that the measurement failed probably due to surface adsorption impurities. Figure 3(b) shows ARPES image of the film after annealing at 400°C for 2 h. The spectrum also does not show any angle dependence. Thus, in this study, we could not obtain good ARPES image of SrFeO₂.

Figure 3(c) shows valence band spectra for the SrFeO₂ films before and after annealing at 200 and 400°C for 2 h. The spectra have peaks at binding energy (E_B) of 5.5 eV and ~8 eV, which agrees well with the previous PES result[5] and the band calculation[4] considering charge-up effects. The peak at $E_B = 5.5$ eV does not change after annealing. On the other hand, the peak intensity at $E_B \sim 8$ eV increases with annealing temperature. This suggests that the surface of the SrFeO₂ films was partially cleaned via ultra-high vacuum annealing method. Thus, it is suggested that it would become possible to obtain clear ARPES image of SrFeO₂ by using higher and longer annealing process.

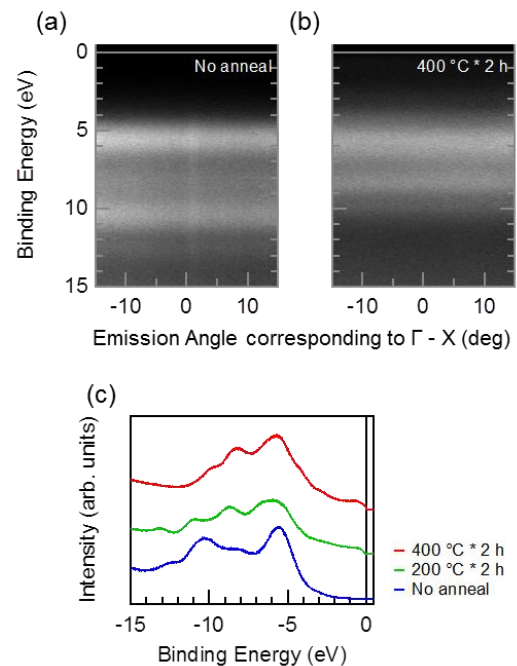


Fig. 3. ARPES images for the SrFeO₂ films (a) before and (b) after annealing at 400°C for 2 h. (c) Valence band spectra for the SrFeO₂ films before and after annealing.

4. Conclusion

We fabricated infinite-layer SrFeO₂ thin films and researched crystal structure stability of the SrFeO₂ films against ultra-high vacuum annealing and performed ARPES on the films. The annealing resulted in partially clean surface of the SrFeO₂ single crystalline film, however, a clear ARPES image was still difficult to be obtained.

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Reference

- [1] Y. Tsujimoto, *et al.*, Nature **450**, 1062 (2007).
- [2] F. D. Romero, *et al.*, Angew. Chem. **126**, 1 (2014).
- [3] T. Katayama, *et al.*, J. Phys. D: Appl. Phys. **47**, 135304 (2014).
- [4] J. M. PrunedaXiang, *et al.*, Phys. Rev. B. **78**, 115101 (2008).
- [5] A. Chikamatsu, *et al.*, J. Electron Spectrosc. Relat. Phenom. **184**, 547 (2012).
- [6] S. Inoue, *et al.*, Appl. Phys. Lett. **92**, 161911(2008).